



Health	2
Fire	1
Reactivity	0
Personal Protection	Ε

Safety Data Sheet Ammonium acetate SDS

Section 1: Chemical Product and Company Identification		
Product Name: Ammonium acetate	Contact Information:	
Catalog Codes: 401		
CAS#: 631-61-8		
RTECS: AF3675000	Email: info@drm-chem.com	
TSCA: TSCA 8(b) inventory: Ammonium acetate	Address: #7, Afshar javan Alley, Sohrevardi	
Synonym: Acetic acid, ammonium salt	St ,Tehran, Iran	
Chemical Name: Ammonium Acetate	post code: 1551818111	
Chemical Formula: CH ₃ COONH ₄ or C ₂ H ₄ O ₂ .NH ₃ or	Tehran Sales: +98 21 88177760	
C ₂ H ₇ NO ₂	Order Online: Drm-chem.com	

Section 2: Composition and Information on Ingredients

Composition:

Name	CAS#	% by Weight
Ammonium acetate	631-61-8	100

Toxicological Data on Ingredients: Ammonium acetate LD50: Not available. LC50: Not available.

Section 3: Hazards Identification

Potential Acute Health Effects:

Hazardous in case of eye contact (irritant), of ingestion. Slightly hazardous in case of skin contact (irritant), of inhalation.

Potential Chronic Health Effects:

CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS: Not available. TERATOGENIC EFFECTS: Not available. DEVELOPMENTAL TOXICITY: Not available. The substance may be toxic to lungs, upper respiratory tract. Repeated or prolonged exposure to the substance can produce target organs damage.

Section 4: First Aid Measures

Eye Contact:

Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Cold water may be used. Get medical attention.

Skin Contact:

Wash with soap and water. Cover the irritated skin with an emollient. Get medical attention if irritation develops. Cold water may be used.

Serious Skin Contact: Not available.

Inhalation:

If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

Serious Inhalation: Not available.

Ingestion:

Do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. If large quantities of this material are swallowed, call a physician immediately. Loosen tight clothing such as a collar, tie, belt or waistband.

Serious Ingestion: Not available.

Section 5: Fire and Explosion Data

Flammability of the Product: May be combustible at high temperature.

Auto-Ignition Temperature: Not available.

Flash Points: Not available.

Flammable Limits: Not available.

Products of Combustion: These products are carbon oxides (CO, CO2), nitrogen oxides (NO, NO2...).

Fire Hazards in Presence of Various Substances: Slightly flammable to flammable in presence of open flames and sparks, of heat. Non-flammable in presence of shocks.

Explosion Hazards in Presence of Various Substances:

Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available.

Fire Fighting Media and Instructions:

SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use water spray, fog or foam. Do not use water jet.

Special Remarks on Fire Hazards: Not available.

Special Remarks on Explosion Hazards: Not available.

Section 6: Accidental Release Measures

Small Spill:

Use appropriate tools to put the spilled solid in a convenient waste disposal container. Finish cleaning by spreading water on the contaminated surface and dispose of according to local and regional authority requirements.

Large Spill:

Use a shovel to put the material into a convenient waste disposal container. Finish cleaning by spreading water on the contaminated surface and allow to evacuate through the sanitary system.

Section 7: Handling and Storage

Precautions:

Keep away from heat. Keep away from sources of ignition. Ground all equipment containing material. Do not breathe dust. Avoid contact with eyes. Wear suitable protective clothing. If you feel unwell, seek medical attention and show the label when possible. Keep away from incompatibles such as oxidizing agents, acids.

Section 8: Exposure Controls/Personal Protection

Engineering Controls:

Use process enclosures, local exhaust ventilation, or other engineering controls to keep airborne levels below recommended exposure limits. If user operations generate dust, fume or mist, use ventilation to keep exposure to airborne contaminants below the exposure limit.

Personal Protection:

Splash goggles. Lab coat. Dust respirator. Be sure to use an approved/certified respirator or equivalent. Gloves.

Personal Protection in Case of a Large Spill:

Splash goggles. Full suit. Dust respirator. Boots. Gloves. A self contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.

Exposure Limits: Not available.

Section 9: Physical and Chemical Properties

Physical state and appearance: Solid. (Deliquescent crystals solid.)
Odor: Acetous (Slight.)
Taste: Not available.
Molecular Weight: 77.08 g/mole
Color: White.
pH (1% soln/water): Not available.
Boiling Point: Not available.
Melting Point: 114°C (237.2°F)
Critical Temperature: Not available.
Specific Gravity: 1.073 [Merck Index]; 1.17 [Handbook of Chemistry and Physcis] (Water = 1)
Vapor Pressure: Not applicable.
Vapor Density: Not available.
Volatility: Not available.
Odor Threshold: Not available.
Water/Oil Dist. Coeff.: Not available.
lonicity (in Water): Not available.
Dispersion Properties: See solubility in water, methanol, acetone.
Solubility: Easily soluble in cold water. Partially soluble in methanol, acetone. Freely soluble in alcohol.

Section 10: Stability and Reactivity Data

Stability: The product is stable.

Instability Temperature: Not available.

Conditions of Instability: Excess heat, incompatible materials

Incompatibility with various substances: Reactive with oxidizing agents, acids.

Corrosivity: Non-corrosive in presence of glass.

Special Remarks on Reactivity:

Deliquescent; Tends to lose ammonia. Incompatible with strong oxidizers such as sodium hyporchlorite, peroxides, potassium chlorate, sodium nitrite.

Special Remarks on Corrosivity: Not available.

Polymerization: Will not occur.

Section 11: Toxicological Information

Routes of Entry: Inhalation. Ingestion.

Toxicity to Animals:

LD50: Not available. LC50: Not available.

Chronic Effects on Humans: May cause damage to the following organs: lungs, upper respiratory tract.

Other Toxic Effects on Humans:

Hazardous in case of ingestion. Slightly hazardous in case of skin contact (irritant), of inhalation.

Special Remarks on Toxicity to Animals:

LD50 [Rat] - Route: Intraperitoneal; Dose: 632/mg/kg. LDL [Mouse] - Route: Intravenous; Dose: 386 mg/kg

Special Remarks on Chronic Effects on Humans: Not available.

Special Remarks on other Toxic Effects on Humans:

Acute Potential Health Effects: Skin: Causes skin irritation. Eyes: Causes eye irritation. Ingestion: May cause irritation of the gastrointestinal tract with generalized discomfort, diarrhea, nausea, vomiting abdominal pain. With ingestion of larges doses of ammonium acetate, there arises the possibility of sufficient absorption to produce diuresis and systemic ammonia poisoning. May also affect behavior/central nervous system(tremor, anxiety, impairment of motor performance or recognition, flaccidity of facial muscles) and respiratory system (dyspnea). Inhalation: Dust can irritate the respiratory tract (nose, throat, lungs) with symptoms of coughing , wheezing,and/or shortness of breath. Chronic Potential Health Effects: Ingestion: May cause some liver dysfunction, and may affect the brain, behavior/central nervous system with symptoms similar to that of acute ingestion. Inhalation: Repeated exposure can irritate the lungs. It may also cause bronchitis to develop with cough, phlegm, and/or shortness of breath.

Section 12: Ecological Information

Ecotoxicity: Not available.

BOD5 and COD: Not available.

Products of Biodegradation:

Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.

Toxicity of the Products of Biodegradation: The product itself and its products of degradation are not toxic.

Special Remarks on the Products of Biodegradation: Not available.

Section 13: Disposal Considerations

Waste Disposal:

Waste must be disposed of in accordance with federal, state and local environmental control regulations.

Section 14: Transport Information

DOT Classification: Not a DOT controlled material (United States).

Identification: Not applicable.

Special Provisions for Transport: Not applicable.

Section 15: Other Regulatory Information

Federal and State Regulations:

Connecticut hazardous material survey.: Ammonium acetate Illinois chemical safety act: Ammonium acetate New York release reporting list: Ammonium acetate Pennsylvania RTK: Ammonium acetate Massachusetts RTK: Ammonium acetate Massachusetts spill list: Ammonium acetate New Jersey: Ammonium acetate New Jersey spill list: Ammonium acetate Louisiana spill reporting: Ammonium acetate California Director's List of Hazardous Substances: Ammonium acetate TSCA 8(b) inventory: Ammonium acetate CERCLA: Hazardous substances.: Ammonium acetate: 5000 lbs. (2268 kg)

Other Regulations: EINECS: This product is on the European Inventory of Existing Commercial Chemical Substances.

Other Classifications:

WHMIS (Canada): Not controlled under WHMIS (Canada).

DSCL (EEC):

R36- Irritating to eyes. S24/25- Avoid contact with skin and eyes. S36/37/39- Wear suitable protective clothing, gloves and eye/face protection.

HMIS (U.S.A.):

Health Hazard: 2

Fire Hazard: 1

Reactivity: 0

Personal Protection: E

National Fire Protection Association (U.S.A.):

Health: 2

Flammability: 1

Reactivity: 0

Specific hazard:

Protective Equipment:

Gloves. Lab coat. Dust respirator. Be sure to use an approved/certified respirator or equivalent. Splash goggles.

Section 16: Other Information

References: Not available.

. Other Special Considerations: Not available

Created: 07/2015

Last Updated: 07/2022

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